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PERFLUOROORGANO TELLURIUM COMPOUNDS: NEW PREPARATIONS AND PROPERTIES OF  $(R_t)_2$ Te AND  $CH_3$ TeR $_t$   $(R_t = CF_3, C_2F_5, C_3F_7, C_6F_5)$ 

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The photochemical reaction of  $(CH_3)_2$ Te with  $CF_3$ I yields  $[(CH_3)_2$ TeCF $_3$ ]I as the main product, which can be isolated in a pure state. The salt thermally decomposes to  $CH_3$ TeCF $_3$  and  $CH_3$ I. The irradiation of a mixture of  $(CH_3)_2$ Te and  $C_2$ F $_5$ I gives  $[(CH_3)_3$ Te]I and  $CH_3$ TeC $_2$ F $_5$ , whereas i-C $_3$ F $_7$ I reacts without irradiation at r.t. to  $[(CH_3)_3$ Te]I and  $CH_3$ Te(i-C $_3$ F $_7$ ). From the same reactions in the presence of equivalent amounts of  $(C_2H_5)_3$ N  $CH_3$ TeR $_f$  are formed in high yields as spectroscopically pure compounds. The main products from the photochemical reactions of  $CH_3$ TeR $_f$  with an excess of R $_f$ I are  $(R_f)_2$ Te and  $(R_f)_2$ Te $_2$ .

Another new method for the preparation of  $(C_2F_5)_2$ Te is the thermal reaction of  $(C_2F_5)_2$ Hg•dipy with TeCl<sub>4</sub>.

The irradiation of a mixture of  $(CH_3)_2$ Te,  $C_6F_5I$  and  $(C_2H_5)_3N$  gives  $(C_6F_5)_2$ Te and  $CH_3$ TeC $_6F_5$ , which can easily be separated, whereas from the thermal reaction of the same mixture  $(C_6F_5)_2$ Te is formed in high yield.

The mechanism of these reactions as well as the n.m.r. and mass spectra and some chemical properties of the partly new compounds will be discussed.